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**Ribofuranosides N-Substituted with meso-Porphyrin as Nucleoside-Like Compounds**

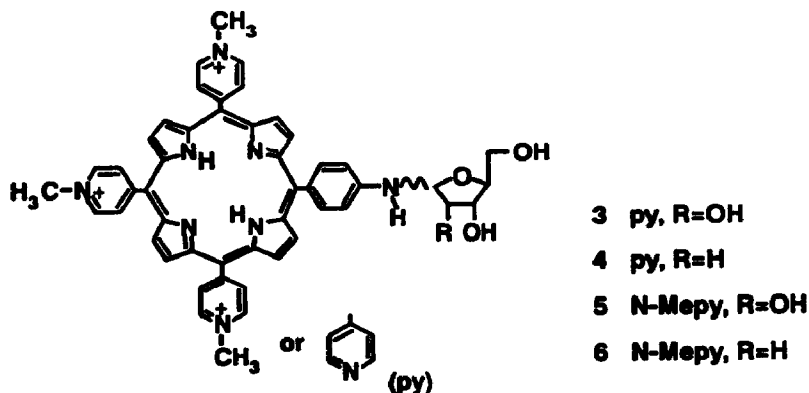
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**Key Words:** ribofuranosides, porphyrinyl-N-glycosides

**Abstract:** Meso-tri(4-pyridyl)porphyrin-p-phenylene-1'-amino-D-ribose and the respective D-2-deoxyribose derivative were synthesized as first representatives of porphyrinyl-N-glycosides.

An earlier report described the synthesis of porphyrins containing O-glycosyl substituents either at all meso-positions or at some  $\beta$ -carbon centers.<sup>1</sup> In this study we describe the synthesis of the first representatives of porphyrinyl-N-glycosides. In these compounds, meso-position is occupied by p-phenylene-1'-aminoribose or p-phenylene-1'-2-deoxyribose. The other three meso-positions contain the 4-pyridyl-or N-methyl-4-pyridinium substituents which are responsible for the solubility of these compounds in water. These porphyrin N-substituted (deoxy)ribofuranosides **3-6**, can be formally considered to be nucleoside-like structures in which a nucleobase is replaced by p-aminophenylporphyrin. Our interest in these compounds represents a continuation of our synthetic efforts toward water-soluble porphyrinyl-nucleosides,<sup>2-4</sup> some of which show tumoricidal activity.<sup>5</sup> The presence of a furanose unit containing the 3'-and 5'-OH groups creates a potential for further synthesis of the analogs of oligonucleotides containing a number of porphyrin units.



The starting porphyrin was the meso-4-acetamidophenyl-tri(4-pyridyl)porphyrin **1** obtained by condensation of pyrrole, pyridine-4-carboxaldehyde and 4-acetanidobenzaldehyde by the Rothmund reaction.<sup>6</sup> The product was isolated as a fourth fraction from a silica gel column,  $\text{CHCl}_3/\text{CH}_3\text{OH}$  60:1 as an eluent. Even with the use of an extremely pure 4-acetamidobenzaldehyde the yield did not exceed 5%. Hydrolysis of acetamido group in  $\text{CF}_3\text{COOH}$  solution by conc. HCl at 80°C for 36 h<sup>7</sup> resulted in the respective aminophenylporphyrin **2** which was purified like **1** with  $\text{CHCl}_3/\text{CH}_3\text{OH}$  80:1 as eluent; yield, 79%. D-ribose

(D-2-deoxyribose) in absolute ethanol solution was then stirred with **2** and anhydrous  $\text{NH}_4\text{Cl}$  at room temperature for 36 h. After removing excess  $\text{NH}_4\text{Cl}$ , solvent and washing out ribose (deoxyribose) with water, the dry product was dissolved in  $\text{CHCl}_3$  and chromatographed using  $\text{CHCl}_3/\text{CH}_3\text{OH}$  40:1 as an eluent. The products were collected as the second fractions with the yields of 64% for meso-tri(4-pyridyl)porphyrin-p-phenylene-1'-amino-D-ribose **3** (37%  $\alpha$  and 63%  $\beta$  anomers) and 54% for the respective D-2-deoxyribose derivative **4** (34%  $\alpha$  and 66%  $\beta$ ). The ratio of anomers was determined for **3** and **4** dissolved in  $\text{H}_2\text{O}/\text{CH}_3\text{CN}$  30:70 by C-18 reverse phase separation, microbore spherisorb S50DS2 as a supporting matrix, G 414 nm. Stirring the DMF solutions of **3** and **4** with excess  $\text{CH}_3\text{I}$  at room temperature for 1 h<sup>8</sup> resulted in N-methylation of 4-pyridyl substituents. The products, **5** and **6**, respectively, were precipitated with diethyl ether from methanol solution; yield 78%.

- <sup>1</sup>HNMR ( $\text{CDCl}_3$ ), ppm: 9.05(m,6H,2,6-py;2H, $\beta$ -pyr),8.85(s,6H, $\beta$ -pyr),8.10(m,6H,3,5-py),8.00(d,2H,2,6-NH-Ph),7.10(d,2H,3,5-NH-Ph),1.95(s,3H,COCH<sub>3</sub>),-2.95(s,2H,porph).
- FAB-MS,(3-nitrobenzyl alcohol matrix),m/z:633 M+; <sup>1</sup>HNMR,200MHz( $\text{CDCl}_3$ ),ppm:9.10(m,6H,2,6-py;2H, $\beta$ -pyr),8.23(m,6H,3,5-py),8.05(d,2H,2,6-NH-Ph),7.15(d,2H,3,5-NH<sub>2</sub>-Ph),-2.85(s,2H,porph). UV-vis ( $\text{CH}_3\text{OH}$ )  $\lambda_{\text{max}}$  nm:646,589,551,514,413(S),247,203.
- FAB-MS,m/z:763(M-2)+; <sup>1</sup>HNMR (DMSO-d<sub>6</sub>), ppm:8.99(m,6H,2,6-py,2H, $\beta$ -pyr),8.80(m,6H, $\beta$ -pyr),8.18(m,6H,3,5-py),7.90(d,2H,2,6-NH-Ph),7.10(d,2H,3,5-NH-Ph),5.05(m,1H,H-1'),4.90(m,2H,H-2',H-3'),4.76(m,1H,H-4'),4.34(m,2H,H-5'),-3.00(s,2H,porph); UV-vis( $\text{CH}_3\text{OH}$ ),  $\lambda_{\text{max}}$  nm:646,589,551,514,414(S),248,209.
- FAB-MS,m/z:747(M-2)+; <sup>1</sup>HNMR(DMSO-d<sub>6</sub>), ppm:9.00(m,6H,2,6-py,2H, $\beta$ -pyr),8.82(m,6H, $\beta$ -pyr),8.21(m,6H,3,5-py),7.92(d,2H,2,6-NH-Ph),7.10(d,2H,3,5-NH-Ph),5.20(m,1H,H-1'),4.70(m,1H,H-3'),4.05(m,1H,H-4'),3.60(m,2H,H-5'),1.95(m,2H,H-2'),-3.00(s,2H,porph); UV-vis( $\text{CH}_3\text{OH}$ ),  $\lambda_{\text{max}}$  nm:646,589,551,514,414(S),252,216.
- <sup>1</sup>HNMR(DMSO-d<sub>6</sub>), ppm:9.45(m,6H,2,6-py),9.00(m,6H,3,5-py;8H, $\beta$ -pyr),7.15(m,4H,NH-Ph),5.85(m,1H,H-1'),5.25(m,2H,H-2',H-3'),4.95(m,1H,H-4'),4.00(m,2H,H-5'),4.65(s,9H,N-CH<sub>3</sub>); UV-vis( $\text{CH}_3\text{OH}$ ),  $\lambda_{\text{max}}$  nm: 653,591,562,519,426(S),253,216.
- <sup>1</sup>HNMR(DMSO-d<sub>6</sub>), ppm:9.42(m,6H,2,6-py),9.10(8H, $\beta$ -pyr),8.95(6H,3,5-py),7.10(m,4H,NH-Ph),5.25(m,1H,H-1'),4.60(s,9H,NCH<sub>3</sub>),4.25(m,2H,H-3',H-4'),3.40(m,2H,H-5'),1.95(m,2H,H-2'), UV-vis( $\text{CH}_3\text{OH}$ ),  $\lambda_{\text{max}}$  nm: 651,593,559,517,424(S),221.

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